

Maharashtra State Board of Technical Education, Mumbai

TEACHING PLAN (TP)

Academic Year: 2025-26 (ODD)

Institute Code and Name: 61303- PPCOE, Karjat

Programme and Code: Electrical Engineering

Course Code: 312308

Name of Faculty: Walunj T.A. /Kharade S.O.

Semester: Second

Course and Code: BSC

Scheme: K

Class: EE2K

COURSE LEVEL LEARNING OUTCOMES (COS)

- CO1 - Identify relevant type of construction materials for the given type of building.
- CO2 - Use the relevant type of special purpose construction materials in the given situation.
- CO3 - Undertake the given type of building construction activity for the given component of building structure.
- CO4 - Design the relevant means of communication for the given building structure.
- CO5 - Use the relevant type of material for finishing purpose in the given situation.

TEACHING-LEARNING & ASSESSMENT SCHEME

Course Code	Course Title	Abbr	Course Category	Learning Scheme				Credits	Paper Duration	Assessment Scheme										Total Marks
				Actual Contact Hrs/Week			SLH			NLH	Theory			Based on LL & TSL Practical				Based on SL		
				C L	T L	L L					FA-TH	SA-TH	Total	FA-PR		SA-PR		SLA		
														Max	Min	Max	Min	Max	Min	
312308	Applied Science	BSC	DSC	4	-	4	8	4	15	30	70*#	100	40	50	20	50@	20			200

Total IKS Hrs for Sem.: 6 Hrs

Abbreviations: CL- Classroom Learning, TL- Tutorial Learning, LL-Laboratory Learning, SLH-Self Learning Hours, NLH-Notional Learning Hours,
 FA - Formative Assessment, SA -Summative assessment, IKS - Indian Knowledge System, SLA - Self Learning Assessment Legends: @ Internal Assessment,
 # External Assessment, *# On Line Examination, @\$ Internal Online Examination

SUGGESTED COS - POS MATRIX FORM

Course Outcomes (COs)	Programme Outcomes (POs)							Programme Specific Outcomes (PSOs)		
	PO-1 Basic and Discipline Specific Knowledge	PO-2 Problem Analysis	PO-3 Design/ Development of Solutions	PO-4 Engineering Tools	PO-5 Engineering Practices for Society, Sustainability and Environment	PO-6 Project Management	PO-7 Life Long Learning	PSO- 1	PSO- 2	PSO- 3
CO1	3	1	1	1	1	1	1			
CO2	3	1	1	1	1	1	1			
CO3	3	2	1	1	1	1	1			
CO4	3	1	-	1	2	2	2			
CO5	3	2	1	2	2	2	2			
CO6	3	1	-	1	2	2	2			

Legends :- High:03, Medium:02, Low:01, No Mapping: - *PSOs are to be formulated at institute level

Maharashtra State Board of Technical Education

K-1

Teaching Plan (TP)

Academic Year: 2025-26

Institute Code: 61303

Program: Civil Engineering, Electrical Engineering

Course Code: 312308

Course: Applied Science (BSC)

Semester: Second (CE-2K/EE-2K)

Name of faculty:

Unit No. (Allocated Hrs.)	CO	TLO	Unit Name and Learning Content Title / Details	No. of Lecture	Plan (From-To)	Actual Execution (From-To)	Teaching method/ Media	Remark
I (9 Hrs)	CO 1		Unit - I Properties of matter and kinematics					
		TLO 1.1	1.1 Deforming Force and Restoring Force, Elasticity, Plasticity, Rigidity.					
		TLO 1.2	1.2 Stress and Strain and their types, elastic limit and Hooke's law, types of moduli of elasticity.					
		TLO 1.3	1.3 Stress -Strain diagram, Poisson's ratio, factors affecting elasticity					
		TLO 1.4	1.4 Newton's laws of motion, and their applications.					
		TLO 1.5	1.5 Angular displacement, angular velocity, angular acceleration, three equations of angular motion, projectile motion, trajectory, range of projectile angle of projection, time of flight					
		TLO 1.6	Work, power and energy: potential energy, kinetic energy, work –energy principle.					
II (10 Hrs)	CO 2		Unit - II Waves and Oscillations					
		TL O 2.1	2.1 Sound waves, amplitude, frequency, time - period, wave-length and velocity of wave, relation between velocity, frequency and time - period of wave.					
		TL O 2.2	2.2 Simple Harmonic Motion , Uniform Circular Motion as Simple Harmonic Motion, Equation of simple harmonic motion , Phase of Simple Harmonic Motion.					
		TL O 2.3	2.3 Resonance , Application of resonance.					
		TL O 2.4	2.4 Resonance concept in prehistoric times, concept of different frequencies (Mantras) used to ignite different chakras in body (IKS).					
		TL O 2.5	2.5 Ultrasonic waves, properties of ultrasonic waves.					

Unit No. (Allocated Hrs.)	CO	TLO	Unit Name and Learning Content Title / Details	No. of Lecture	Plan (From-To)	Actual Execution (From-To)	Teaching method/ Media	Remark
III (11 Hrs)		TL O 2.6	2.6 Piezoelectric and Magnetostriction method to produce ultrasonic waves .					
		TL O 2.7	2.7 Applications of ultrasonic waves.					
	CO 3		Unit - III Modern Physics (Photoelectricity , X rays, LASER and nanotechnology)					
		TL O 3.1	3.1 Planck's hypothesis, properties of photons.					
		TL O 3.2	3.2 Photo electric effect: threshold frequency, threshold wavelength, stopping potential, Work function, characteristics of photoelectric effect, Einstein's photoelectric equation					
		TL O 3.3	3.3 Photoelectric cell and LDR : principle ,Working and applications					
		TL O 3.4	3.4 Production of X-rays by modern Coolidge tube, properties and engineering applications.					
		TL O 3.5	3.5 Laser: properties, absorption, spontaneous and stimulated emission,					
		TL O 3.6	3.6 Population inversion, active medium, optical pumping, three energy level system, He-Ne Laser.					
		TL O 3.7	3.7 Engineering applications of Laser.					
		TL O 3.78	3.8 Nanotechnology : Properties of nanomaterials (optical, magnetic and dielectric properties) , applications of nanomaterials, Metallic Bhasma (Ancient Ayurveda, IKS).					
IV (10 Hrs)	CO 4		Unit - IV Metals and Alloys					
		TL O 4.1	4.1 Ancient Indian Metallurgy (IKS)					
		TL O 4.2	4.2 Metals: Occurrence of metals in free and combined state. Basic concepts : Mineral, ore, gangue, flux and slag, metallurgy.					
		TL O 4.3	4.3 Metallurgy:Extraction processes of metal from ore Concentration : Gravity separation, electromagnetic separation, froth floatation, calcination and roasting, Reduction : Smelting, aluminothermic process, Refining, poling , electrorefining.					

Unit No. (Allocated Hrs.)	CO	TLO	Unit Name and Learning Content Title / Details	No. of Lecture	Plan (From-To)	Actual Execution (From-To)	Teaching method/ Media	Remark
		TL O 4.4	4.4 Mechanical properties of metals :Hardness, ductility, malleability, tensile strength, toughness, machinability, weldability, forging, soldering, brazing, castability.					
		TL O 4.5	4.5 Alloys: Purposes of making alloys with examples.					
		TL O 4.6	4.6 Preparation methods of alloys : Fusion, compression.					
		TL O 4.7	4.7 Classification of alloys :Ferrous and non-ferrous alloys Ferrous alloys: Composition ,properties and applications of low carbon, medium carbon, high carbon steels. Non- ferrous alloy:Composition ,properties and applications of Brass, Bronze, Duralumin, Tinman Solder, Woods metal.					
V (8 Hrs)	CO 5		Unit - V Water Treatment					
		TL O 5.1	5.1 Hard and soft water, causes of hardness, types of hardness					
		TL O 5.2	5.2 Hard water in boilers and prevention: Boiler corrosion, caustic embrittlement, priming and foaming, scales and sludges,and methods of prevention of boiler corrosion.					
		TL O 5.3	5.3 Methods of water softening: lime soda process (hot lime soda and cold lime soda process), zeolite process, ion exchange process.					
		TL O 5.4	5.4 Potable water treatment: Sedimentation, coagulation, filtration and sterilization .					
		TL O 5.5	5.5 Wastewater treatment: Sewage treatment, BOD and COD of sewage water.					
		TL O 5.6	5.6 pH and pOH: Concept of pH, pOH, pH Scale, Numerical.					
VI (12 Hrs)	CO 6		Unit - VI Fuels and Combustion					
		TL O 6.1	6.1 Fuel: Calorific value and ignition temperature, classification.					
		TL O 6.2	6.2 Solid fuels: Coal, Classification and composition , Proximate analysis, Ultimate analysis, Calorific value of coal by Bomb calorimeter.					
		TL O 6.3	6.3 Liquid fuels: Fractional distillation of crude petroleum, boiling range, composition, propertie Knocking,					

<i>Unit No. (Allocated Hrs.)</i>	<i>CO</i>	<i>TLO</i>	<i>Unit Name and Learning Content Title / Details</i>	<i>No. of Lecture</i>	<i>Plan (From-To)</i>	<i>Actual Execution (From-To)</i>	<i>Teaching method/ Media</i>	<i>Remark</i>
			cracking, octane number and cetane number.					
		TL O 6.4	6.4 Gaseous fuels: Biogas, LPG, and CNG. Combustion equation of gaseous fuels, mass and volume of air required for complete combustion.					
		TL O 6.5	6.5 Green hydrogen: Producing green hydrogen by electrolysis from renewable sources , Advantages and disadvantages of green hydrogen.					
		TL O 6.6	6.6 Electrical conductance in metals and electrolytes, specific conductance, equivalent conductance, cell constant					
		TL O 6.7	6.7 Cells and batteries :Construction ,working and applications of dry cell, lead acid storage cell H ₂ - O ₂ fuel cell, Ni-Cd battery and Lithium ion battery					

X. ASSESSMENT METHODOLOGIES/TOOLS

➤ Formative assessment (Assessment for Learning)

- Tests
- Rubrics for COs Assignment
- Midterm Exam
- Self-learning
- Term Work
- Seminar/Presentation

➤ Summative Assessment (Assessment of Learning)

- End Term Exam Theory
- Micro-project
- Tutorial Performance

(Name & Signature of Staff)

(Name & Signature of HOD)